Chapter 4

Biodiesel Feedstock and Production Technologies: Successes, Challenges and Prospects

Y.M. Sani, W.M.A.W. Daud and A.R. Abdul Aziz

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1. Introduction

In order to achieve the biodiesel central policy of protecting the environment, replacing petroleum diesel and protecting and/or creating jobs, a good understanding of biodiesel history is essential. This is because consumers always tend to buy cheap rather than “green” fuels. Moreover, it is more difficult for a new technology to dislodge one that has reached societal standard. The more the popular technology is used, the more it improves; becoming less expensive due to wider market potentials. Petrodiesel has become the “life-blood” of our economy. It would be almost impossible to find a commercial product today that does not consume diesel fuel during its production and distribution [1-4]. Therefore, the aim of this chapter is to provide an overview on the history and motivation, successes, challenges and prospects of biodiesel as source of energy. This will provide a global outlook in making biodiesel an economical and eco-friendly alternative to petroleum diesel.

The historical developments of the biofuel industry in general and biodiesel in particular, is unlike many industries. This is because the driving factors for its advances are more of economics and politics than technological [5]. As early as 1853, transesterification was conducted on vegetable oil in the search for a cheap method to produce glycerine for producing explosives during World War II by E. Duffy and J. Patrick [6-8]. In 1937, G. Chavanne, Belgian scientist patented the “Procedure for the transformation of vegetable oils for their uses as fuels”. “Biodiesel” as a concept was thus established [9]. It is a simple process where alkoxy group of an ester compound (oil or fat) is exchanged with an alcohol. However, it was not until 1977 that first patent on commercial biodiesel production process was applied for by Expedito Parente; a Brazilian scientist [10].

Prior to the discovery of and boom in fossil fuels, power was mainly generated from steam. However, the use of hydro-energy consumes large resources coupled with the inefficiencies
of the steam engine where only about 10 to 12% efficiency is derived from new power
generation plant. A patent for an efficient thermal engine which was to be operated on peanut
oil was filed in 1892 by Rudolph Diesel in Germany. By 1893, Diesel’s invention was demon-
strated in an exhibition in Paris. Within five years of its invention, Diesel’s engine ran on its
own power with 75% efficiency against its initial 26% efficiency [11]. In 1912, Diesel published
two articles [12,13] in which he reflected:

“The fact that fat oils from vegetable sources can be used may seem insignificant to-day, but such oils may perhaps become in course
of time of the same importance as some natural mineral oils and the tar products are now. (...) In any case, they make it certain that
motor power can still be produced from the heat of the sun, which is always available for agricultural purposes, even when all our
natural stores of solid and liquid fuels are exhausted.”

The demand for biofuels began to increase in America from the 1890’s to 1920’s. These were
attributed to the pioneering efforts on the diesel engine by Adolphus Busch and Clessie L.
Cummins along with other engine manufacturers. However, the biofuel industry was faced
with a major challenge of cheap and readily available feedstock. Unfortunately for the biofuel
industry, at this same period, the petroleum industries found out more advanced technologies
for improving the properties of the “black gold”. The discoveries of large reservoirs and
developments created new markets for this “black gold”. Therefore, by 1940, diesel engines
were altered to enable them use petroleum-based fuels which have lower viscosities. There-
after, the sales of biodiesel were weakened and the production structure was pushed to the
background. Therefore, no significant efforts were made to increase the public awareness on
its potentials. This period witnessed increased demands for automobiles which were propelled
by petroleum fuels. The availability of public funds, and new transportation infrastructure
such as interstate and highway systems helped in this regard [14].

The early post-WWII fossil fuel demand and supply was influenced by the commencement of
offshore oil and gas production in 1945 at the Gulf of Mexico and the invention of jet aircraft
[14]. However in the 1970s, speculations regarding the finite nature of the fossil oil reserves
became an issue worth pondering over. In 1973 and 1978, OPEC reduced oil supplies and
increased the prices to meet with the shortages of the petroleum crisis of that time. This marked
the reemergence of the potentials of biofuels in the public consciousness. Thus in 1979, South
Africa started the commercial development of biodiesel. Sunflower oil was transesterified and
refined to a standard similar to petroleum diesel fuel [15]. The outcome was the discovery of
several sources and technologies that improved engine performance with reduced environ-
mental impacts. Experiences from past were used in achieving improved efficiencies, while
reducing costs by developing the renewable energy marketing advantage.

The procedure for the production, quality and engine-testing for biodiesel was finalized and
published internationally in 1983. The South African technology was obtained by Gaskoks; an
Austrian company. Gaskos established the first pilot plant for biodiesel production in 1987.
By April of 1989, the firm set up the first commercial-scale plant producing 20 million gallon per year (MGPY). However during this period, biodiesel was only being produced on a noncommercial scale in the United States. The growth in producing biodiesel in Europe began in 1991 because of the need to reduce environmental impacts from emissions of greenhouse gases (GHG). Three years later, the first commercial biodiesel production was started in America. By 2000, the Commodity Credit Corporation started subsidizing value-added agriculture towards biodiesel production. The past decade (2002 to 2012) witnessed an unprecedented production of biodiesel. Incentives from policy makers such as tax exemptions, tax credits and renewable fuel standards aided the biodiesel growth. However, some properties of biodiesel also contributed to the unprecedented growth we are witnessing in the biodiesel industry [16-18].

The increasing interests on biodiesel is fueled by the need to find a sustainable diesel fuel alternative. This is mainly because of environmental issues, apprehensions over energy independence and skyrocketing prices. Several processing options are available for the biodiesel production. The various feedstocks and processing conditions provide several processing technologies. The choice of a particular technology is dependent on catalyst and the source, type and quality of feedstock. Others include postproduction steps such as product separation and purification and catalyst and alcohol recovery. The dominant factor in the production process is the cost of feedstock while capital costs contribute only about 7%. It is therefore essential to utilize cheap feedstock to reduce the overall production costs. In the same regards, some technologies are designed to handle variety of feedstocks.

2. Past achievements

Non-fossil fuel alternatives are favored because of their common availability, renewability, sustainability, biodegradability, job creation, regional development and reduced environmental impacts. Table 1 summarizes some of the major successes of biodiesel.

2.1. Feedstocks

Numerous feedstocks have been experimented in biodiesel production. Advancements from such experimentations led to establishment of waste-to-wealth biodiesel production. Cheap and readily available raw materials such as used cooking oil and yellow grease are used for producing biodiesel. These efforts helped in reducing the environmental impacts associated with dumping in landfills as well as saves the cost of paying for such dumping. Another notable success is the use of Jatropha or the “miracle plant” in many developing countries. The fact that it can be cultivated almost anywhere with minimal irrigation and less intensive care, made it suitable for peasant farmers. Sustained high yields were obtained throughout its average life cycle of 30–50 years. Castor plantation are also intercropped with jatropha to improve the economic viability of jatropha within the first 2 to 3 years [19]. Another oil crop that is used to improve soil quality is the nitrogen-fixing Pongamia pinnata. It produces seeds with significant oil contents.
2.2. Technologies

Biodiesel is one of the most thoroughly tested alternative fuel in the market today. Studies by many researchers have confirmed similar engine performance of biodiesel to petroleum diesel. Transesterification produce oil with similar brake power as obtained with diesel fuel. Minimal carbon deposits were noticed inside the engine except the intake valve deposits which were slightly higher. The level of injector coking was also reduced significantly lower than that observed with D2 fuel [7,17]. An important breakthrough in transesterification is the Mcgyan Process®, which can utilize various inexpensive, non-food-grade and free fatty acids (FFAs) containing feedstocks (Figure 1). The process can be small in physical size and it utilizes heterogeneous catalysts to produce biodiesel within 4 s [20,21]. The easy fatty acid removal or EFAR system ensures that no wastes are produced from the process. It eliminates post production costs such as the washing and neutralization steps. To achieve 100% conversion, it recycles all unreacted feedstock and excess alcohol back into the reactor. Energy efficiency is also achieved through heat transfer mechanism; in-coming cold reactants are preheated by the out-going hot products [20,21].

![Figure 1. Process flow diagram of a biodiesel plant based on the Mcgyan process[21].](image-url)
<table>
<thead>
<tr>
<th>Economic &amp; social impact</th>
<th>Environment impact</th>
<th>Energy security</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sustainability; made from agricultural or waste resources</td>
<td>Reduced 78% GHG emissions</td>
<td>Reduced dependence on fossil fuels</td>
</tr>
<tr>
<td>Fuel diversity &amp; improved fuel efficiency &amp; economy</td>
<td>Reduced air pollution</td>
<td>Domestic targets</td>
</tr>
<tr>
<td>Improved rural economy</td>
<td>Biodegradability</td>
<td>Supply reliability</td>
</tr>
<tr>
<td>Increased income tax &amp; trade balances</td>
<td>Improved land &amp; water use</td>
<td>Readily available</td>
</tr>
<tr>
<td>International competitiveness</td>
<td>Carbon sequestration</td>
<td>Renewable</td>
</tr>
<tr>
<td>Increased investments on feedstocks &amp; equipment</td>
<td>Lower sulfur content</td>
<td>Domestic distribution</td>
</tr>
<tr>
<td>Technological developments (R &amp; D)</td>
<td>Lower aromatic content</td>
<td>Improved fuel economy</td>
</tr>
<tr>
<td>Higher cetane number (52 vs. 48), lubricity &amp; flash point</td>
<td>Lesser toxicity</td>
<td>Comparable energy content (92.19%)</td>
</tr>
<tr>
<td>Knowledge development &amp; diffusion</td>
<td>Safer handling &amp; storage</td>
<td>Strict quality requirements are met</td>
</tr>
<tr>
<td>Strong growth in demand &amp; market formation</td>
<td>Viscosity 1.3 to 1.6 times that of D2 fuel</td>
<td></td>
</tr>
<tr>
<td>Improved engine performance</td>
<td>Good energy balance (3.24:1 vs. 0.88:1)</td>
<td></td>
</tr>
<tr>
<td>Reduces the need for maintenance &amp; prolongs engine life</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Compatible with all conventional diesel engines</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Offers the same engine durability &amp; performance</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Has the potential of displacing petroleum diesel fuel</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Comparable start-up, torque range &amp; haulage rates</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Table 1. Major achievements of biodiesel [16,23-27]

2.3. Environmental impacts and health effects

A 78% reduction in GHG emission was reported by the U.S. Departments of Agriculture and Energy with biodiesel usage. Essentially, biodiesel is non-aromatic and sulphur-free as compared with petrodiesel which contains 20 to 40 wt.% aromatic compounds and 500 ppm
SO₂ [7]. The potential of pure biodiesel to form ozone (smog) from hydrocarbons is 50% less. Also, sulfates and oxides of sulfur (major constituents of acid rain) are essentially eliminated from the exhaust emissions compared to petrodiesel. These help in curbing the increasing global warming problems. Average decrease of 22.5% for smoke density, 17.1% for CO and 14% for CO₂ have been reported when biodiesel was used [22]. Human life expectancy is thereby enhanced because of improved air quality.

2.3.1. Energy independence

Biodiesel reduces the excessive reliance on fossil fuels. This enhances the global energy security [17]. It also has the potential to replace oil importation since it is produced domestically, thereby providing additional market for agricultural products. It supports the rural communities where it is cultivated by protecting and generating jobs. Producing biofuels equivalent to 1% of automobile fuel consumption in the EU protected and/or created approximately 75,000 jobs [16]. Approximately, for every unit of fossil energy used in biodiesel production, 4.5 units of energy is gained. Moreover, lesser energy is required for biodiesel production than the energy derived from the final product [16].

3. Different feedstocks used in the production of biodiesel

More than 350 oil-bearing crops have been identified as potential sources for producing biodiesel. However, only palm, jatropha, rapeseed, soybean, sunflower, cottonseed, safflower, and peanut oils are considered as viable feedstocks for commercial production [28].

3.1. Edible feedstocks

Depending on availability, different edible oils are utilized as feedstocks for biodiesel production by different countries. Palm oil and coconut oil are commonly used in Malaysia and Indonesia. Soybean oil is majorly used in U.S. [30].

3.2. Non-edible feedstocks

In order to reduce production costs and to avoid the food-for-fuel conflict, inedible oils are used as the major sources for biodiesel production. Compared to edible oils, inedible oils are affordable and readily available. They are obtained from Jatropha curcas (jatropha or ratanjyote or seemaiakattamankku), Pongamia pinnata (karanja or honge), Calophyllum inophyllum (nag-champa), Hexca brasiliensis (rubber seed tree), Azadirachta indica (neem), Madhuca indica and Madhuca longifolia (mahua), Ceiba pentandra (silk cotton tree), Simmondsia chinensis (jojoba), Euphorbia tirucalli, babassu tree, microalgae, etc. [31]. Among the 75 plant species which have more than 29% oil in their seed/kernel; palm, Jatropha curcas, and Pongamia pinnata (Karanja) were found to be the most suitable for biodiesel production [32]. Many European countries utilize rapeseed [29]. During World War II, oil from Jatropha seeds was used as blends with and substituted for diesel [33,34]. It has been reported that biodiesel produced from palm and
Jatropha have physical properties in the right balance; conferring it with adequate oxidation stability and cold performance [35]. Most of the strict requirements set by the American and European biodiesel standards for biodiesel have been achieved [36]. The major oils used for producing biodiesel are presented in Table 2.

<table>
<thead>
<tr>
<th>Group</th>
<th>Source of oil</th>
</tr>
</thead>
<tbody>
<tr>
<td>Major oils</td>
<td>Coconut (copra), corn (maize), cottonseed, canola (a variety of rapeseed), olive, peanut (groundnut), safflower, sesame, soybean, and sunflower.</td>
</tr>
<tr>
<td>Nut oils</td>
<td>Almond, cashew, hazelnut, macadamia, pecan, pistachio and walnut.</td>
</tr>
<tr>
<td>Other edible oils</td>
<td>Amaranth, apricot, argan, artichoke, avocado, babassu, bay laurel, beech nut, ben, Borneo tallow nut, carob pod (algaroba), cohune, coriander seed, false flax, grape seed, hemp, kapok seed, lallemandia, lemon seed, macaua fruit (Acrocomia sclerocarpa), meadowfoam seed, mustard, okra seed (hibiscus seed), perilla seed, pequi, (Caryocar brasiliensis seed), pine nut, poppy seed, prune kernel, quinoa, ramtil (Guizotia abyssinica seed or Niger pea), rice bran, tallow, tea (camellia), thistle (Silybum marianum seed), and wheat germ.</td>
</tr>
<tr>
<td>Inedible oils</td>
<td>Algae, babassu tree, copaiba, honge, jatropha or ratanjyote, jojoba, karanja or honge, mahua, milk bush, nagchampa, neem, petroleum nut, rubber seed tree, silk cotton tree, and tall.</td>
</tr>
<tr>
<td>Other oils</td>
<td>Castor, radish, and tung.</td>
</tr>
</tbody>
</table>

Table 2. Major oil species for biodiesel production [37]

3.2.1. Algae oil

Currently, algae-based biodiesel is the focus of many research interests because they have the potential to provide sufficient oil for global consumption. It has the potential to produce biodiesel yields >100 times those attainable per hectare from plant feedstock (Table 3). Besides their high lipid contents and fast growth rate, microalgae have the potential to mitigate the competitions for land-use and food-for-fuel conflicts. They are also able to reduce the GHG effect via CO₂ sequestration [38]. Microalgae can be cultivated in habitats which are not favorable for energy crops. Compared with oilseeds, the harvesting and transportation costs of microalgae are relatively low. *Nannochloropsis*, members of the marine green algae are considered the most suitable candidates for biodiesel production. These strains have shown high lipid content and biomass productivity. However, research in this area especially algal oil extraction is still limited and in early stages.

3.2.2. Other feedstocks

Used vegetable oils (UCO), yellow grease (8-12 wt% FFA), brown grease (>35 wt% FFA), and soapstock (by-product of refining vegetable oils) are potential feedstocks for biodiesel production. Their low costs and availability make them suitable for reducing the production
costs of biodiesel. To achieve this however, the problems associated with high FFA which are common to these feedstocks, particularly when alkaline catalysts are employed need attention. Solid acid catalysts are currently receiving great attention because they are suitable for feedstocks containing FFAs [39-41]. Another process that has the potential of processing these feedstocks is supercritical transesterification. The pretreatment step, soap and catalyst removal common to alkaline catalysis are eliminated since the process requires no catalyst [42,43]. The process has fast reaction rate which significantly reduces the reaction time [44]. The process is insensitive to water and FFAs [43,45]. However, this method is not economical because it requires high reaction temperature, pressure and higher molar ratio of alcohol to feedstock [42,43,46]. Another interesting feedstock is Salicornia bigelovii (Halophytessuch). It can produce equal biodiesel yields obtained from soybeans and other oilseeds. They grow in saltwater of coastal areas unsuitable for energy crops.

<table>
<thead>
<tr>
<th>Microalgae/Plant</th>
<th>Oil yield (L/ha/year)</th>
<th>Oil content (% wt in biomass)</th>
<th>Required land (M ha⁻¹)</th>
<th>Biodiesel productivity (kg biodiesel/ha/year)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Microalgae² (high oil content)</td>
<td>136 900</td>
<td>70</td>
<td>2</td>
<td>121 104</td>
</tr>
<tr>
<td>Microalgae³ (low to low oil content)</td>
<td>58 700 to 97 800</td>
<td>30 to 50</td>
<td>4.5</td>
<td>51 927-85 515</td>
</tr>
<tr>
<td>Oil palm (Elaeis guineensis)</td>
<td>5 950</td>
<td>30 to 60</td>
<td>45</td>
<td>4747</td>
</tr>
<tr>
<td>Jatropha (Jatropha curcas L.)</td>
<td>1 892</td>
<td>Kernel: 50 to 60 Seed: 35 to 40</td>
<td>140</td>
<td>656</td>
</tr>
<tr>
<td>Canola/Rapeseed (Brassica napus L.)</td>
<td>1 190</td>
<td>38 to 46</td>
<td>223</td>
<td>862</td>
</tr>
<tr>
<td>Soybean (Glycine max L.)</td>
<td>446</td>
<td>15 to 20</td>
<td>594</td>
<td>562</td>
</tr>
<tr>
<td>Corn/Maize (Germ) (Zea mays L.)</td>
<td>172</td>
<td>44 to 48</td>
<td>1540</td>
<td>152</td>
</tr>
</tbody>
</table>

Table 3. Estimated oil content, yields and land requirement for various biodiesel feedstocks.[36,47,48]

4. Methods of oil extraction

The three common methods used in extracting oil are: (i) Mechanical extraction, (ii) solvent extraction and (iii) enzymatic extraction.
4.1. Mechanical extraction method

This method is used by smaller production firms for processing less than 100,000 kg/day. Usually, an engine driven screw press or a manual ram press is used to extract 68–80% or 60–65% of the available oil respectively. Pretreatment such as dehulling and cooking increase oil yields to 89% and 91% after single and dual pass respectively [48,49]. However, most of the mechanical presses are designed for particular seeds which affect yields with other seeds. Also, extra treatments such as degumming and filtration are required for oil extracted by this technique.

4.2. Chemical (solvent) extraction method

The commonly used chemical methods are: (1) soxhlet extraction, (2) Ultrasonication technique and (3) hot water extraction [48,49]. Solvent extraction (or leaching) is typically used for processing more than 300,000 kg/day [50]. Yields are affected by particle size, solvent type and concentration, temperature and agitation. To increase the exposure of the oil to the solvent, the oilseeds are usually flaked. After extraction, the oil-solvent mixture or miscella, is filtered while heat is used to vaporize the solvent from the miscella. Steam is injected to remove any solvent remaining from the oil. The immiscibility of the solvent and steam vapors is used to separate them in a settling tank after condensation. The highest oil yields are obtained with n-hexane. However, the process requires higher energy and longer time compared to other methods. Furthermore, the human health and environmental impacts associated with toxic solvents, waste water generation and emissions of volatile organic compounds are challenges facing this method.

4.3. Enzymatic extraction method

Oilseeds are reduced to small particles and the oil is extracted by suitable enzymes. Volatile organic compounds are not produced by this method which makes it environmentally friendly when compared to the other methods. However, it has the disadvantage of long processing time and high cost of purchasing enzymes [51].

5. Technologies used in biodiesel production

Several researches were carried out to overcome or minimize the problems associated with producing biodiesel. The methods that have been used for minimizing the viscosity of vegetable oils for practical application in internal combustion engines include: pyrolysis, microemulsification, blending (diluting) and transesterification. Dilution and microemulsification are not production processes and are therefore not discussed in this chapter. A summary of vegetable oils and animal fats and the major biodiesel production technologies are presented in Table 4.
5.1. Pyrolysis or catalytic cracking

Pyrolysis is the heating of organic matter in the absence of air to produce gas, a liquid and a solid [52]. Heat or a combination of heat and catalyst is used to break vegetable oils or animal fats into smaller constituents. Olefins and paraffins are thus obtained with similar properties to petrodiesel where such products derived the name “diesel-like-fuel” [53]. Studies on effects of rapeseed particle size showed that the product yield is independent of the oilseed particle size [52]. The maximum temperature range for conversion of bio-oil is 400°C to 450°C [54]. Rapid devolatilization of cellulose and hemicellulose occur at this temperature. Heating rate and temperature have significant effects on bio-oil yields, char and gas released from olive [55]. The viscosity, flash and pour points and equivalent calorific values of the oil are lower than diesel fuel. Though the pyrolyzate has increased cetane number, it is however lower than that of diesel oil. Apart from reducing the viscosity of the vegetable oil, pyrolysis enables decoupling of the unit operation equipment in shorter time, place and scale. It produces clean liquids which needs no additional washing, drying or filtering. Product of pyrolysis consists of heterogeneous molecules such as water, particulate matter, sulfur, alkanes, alkenes and carboxylic acids [39,56]. Consequently, it is difficult to characterize fuel obtained from pyrolysis [52]. This process is energy consuming and needs expensive distillation unit. Moreover, the sulfur and ash contents make it less eco-friendly [57].

5.2. Transesterification (alcoholysis)

Transesterification is the most widely employed process for commercial production of biodiesel. It involves heating the oil to a designated temperature with alcohol and a catalyst, thereby restructuring its chemical structure. This conversion reduces the high viscosity of the oils and fats. For the transesterification of triglyceride (TG) molecule, three consecutive reactions are needed. In these reactions, FFA is neutralized by the TG from the alcohol. One mole of glycerol and three moles of alkyl esters are produced (for each mole of TG converted) at the completion of the net reaction. These separate into three layers, with glycerol at the bottom, a middle layer of soapy substance, and biodiesel on top [57]. Transesterification is a reversible reaction. To obtain reasonable conversion rates therefore, it requires a catalyst. The reaction conditions, feedstock compositional limits and post-separation requirements are predetermined by the nature of the catalyst. Table 5 presents a general overview of the several transesterification techniques for biodiesel production.

5.2.1. Homogeneous alkali-catalyzed transesterification

Alkali catalysts such as NaOH and KOH were preferred over other catalysts because of their ability to enhance faster reaction rates [63]. This is because they are readily available at affordable prices and enable fast reaction rates [24]. Detailed review on base-catalyzed transesterification of vegetable oils can be found in ref [64]. However, homogeneous catalysis has been faced with the been faced with the problems saponification, highly sensitive to FFAs, expensive separation requirement, waste water generation and high energy consumption.
5.2.2. Homogeneous acid-catalyzed transesterification

Though the performance of this method is not strongly affected by FFAs in the feedstock, the process is not as popular as the base-catalyzed process. This is because the use of strong acids such as H₂SO₄ [65,66], HCl, BF₃, H₃PO₄, and organic sulfonic acids [67], is associated with higher costs and environmental impacts. Moreover, the technique is about 4000 times slower than the homogeneous base-catalyzed reaction. The mechanism of the acid-catalyzed transesterification can be found in ref [68].

5.2.3. Heterogeneous acid and base-catalyzed transesterification

Solid acid can simultaneously catalyze the esterification and transesterification without the need for pretreating feedstocks with high FFAs. Thus, this technique has the potential of reducing the high cost of biodiesel production by directly producing biodiesel from readily available and low-cost feedstocks [67].

Solid basic catalysts also have the potential of reducing the cost of biodiesel production because of lesser catalyst consumption, reuse and regeneration. However, these catalysts have some disadvantages which hinder their wide acceptability. These include mass transfer (diffusion) problem which reduces the rate of reaction as a result of the formation of three phases with alcohol and oil. Other problems associated with base catalyzed transesterification are loss of catalyst activity in the presence of water and post-production costs such as product separation, purification and polishing.

5.2.4. Enzymatic transesterification

Some of the problems associated with homogeneous catalysts such as expensive product separation, wastewater generation, and the presence of side reactions are avoided with enzymatic transesterification [69]. Enzyme immobilization is usually done to enhance the product quality, increase the number of times the catalyst is reused and to reduce cost [28,70]. However, several technical difficulties such as high cost of purchasing enzymes, product contamination, and residual enzymatic activity are limiting the applicability of this technique.

5.2.5. Supercritical alcohol transesterification

Unlike the conventional transesterification of two heterogeneous liquid phases involving alcohol (polar molecule) and non-polar molecules (TGs), supercritical transesterification is done in single homogeneous phase. Subjecting solvents containing hydroxyl groups (such as water and alcohol) to conditions in excess of their critical points make them to act as superacids. Under supercritical conditions, alcohol serves a dual purpose of acid catalyst and a reactant [46,71]. The absence of interphase solves the mass transfer limitations which gives the possibility of completing the reaction in minutes rather than several hours. In fact, the Mcgyan Process® was used to produce biodiesel under 4 s [19,20]. However, this process is not economical especially for commercial production as it requires expensive reacting equipment due to high temperature and pressure [72]. Studies are currently being undertaken in order to reduce these high reacting conditions.
<table>
<thead>
<tr>
<th>Direct use</th>
<th>Dilution with vegetable oils</th>
<th>Microemulsion of oils</th>
<th>Pyrolysis and catalytic cracking</th>
<th>Transesterification of oils and fats</th>
</tr>
</thead>
<tbody>
<tr>
<td>Advantages</td>
<td>Advantages</td>
<td>Advantages</td>
<td>Advantages</td>
<td>Catalytic</td>
</tr>
<tr>
<td>Simple process</td>
<td>Simple process and non-polluting</td>
<td>Simple process and non-polluting</td>
<td>Simple process &amp; non-polluting</td>
<td>Acid-catalyzed</td>
</tr>
<tr>
<td>Disadvantages</td>
<td>Disadvantages</td>
<td>Disadvantages</td>
<td>Disadvantages</td>
<td>Alkali catalytic</td>
</tr>
<tr>
<td>Highly viscous</td>
<td>Highly viscous</td>
<td>Incomplete combustion</td>
<td>Contains heterogeneous molecules</td>
<td>Enzyme-catalyzed</td>
</tr>
<tr>
<td>Highly unstable</td>
<td>Highly unstable</td>
<td>Injector needle sticking</td>
<td>Low purity</td>
<td></td>
</tr>
<tr>
<td>Low volatility</td>
<td>Low volatility</td>
<td>Carbon deposits</td>
<td>Requires high temperature</td>
<td>Catalytic supercritical alcohol</td>
</tr>
<tr>
<td>Not suitable for commercial production</td>
<td>Not suitable for commercial production</td>
<td>Not suitable for commercial production</td>
<td>Requires expensive equipment</td>
<td>See Table 5 for advantages and disadvantages</td>
</tr>
</tbody>
</table>

Table 4. Use of vegetable oils and animal fats and major biodiesel production processes.

5.3. Technologies

5.3.1. Microwave assisted transesterification

The microwave irradiation as energy stimulant has been attracting the attention of many researchers. This is because the reaction process fast (within minutes), it employs a lower alcohol-oil ratio and it reduces by-products quantities. It uses a continuously changing electrical and magnetic fields to activate the smallest degree of variance of the reacting molecules. These rapidly rotating charged ions interact easily with minimal diffusion limitation [73]. However, this process also has commercial scale-up problem because of high operating conditions and safety aspects [74]. An even more daunting challenge is in increasing the irradiation penetration depth beyond a few centimeters into the reacting molecules.

5.3.2. Ultrasound assisted transesterification

This process utilizes sound energy at a frequency beyond human hearing. It stretches and compresses the reacting molecules in an alternating manner. Application of high negative
pressure gradient beyond the critical molecular distance forms cavitation bubbles. Some of the bubbles expand suddenly to unstable sizes and collapse violently. This causes emulsification and fast reaction rates with high yields since the phase boundary has been disrupted [75-77].

<table>
<thead>
<tr>
<th>Chemical catalysed</th>
<th>Chemical catalysed (Modified)</th>
<th>Biochemical catalysed</th>
<th>Noncatalysed methanol</th>
</tr>
</thead>
<tbody>
<tr>
<td>Homogeneous Acid</td>
<td>Homogeneous base</td>
<td>Heterogeneous Acid</td>
<td>Microwave ir-</td>
</tr>
<tr>
<td>Merits</td>
<td>Merits</td>
<td>Merits</td>
<td>irradiation</td>
</tr>
<tr>
<td>Employs feedstocks</td>
<td>Reaction is 4000 times faster</td>
<td>High possibility of</td>
<td>High possibility of</td>
</tr>
<tr>
<td></td>
<td>than homogeneous acid</td>
<td>reusing and re-</td>
<td>reusing and re-</td>
</tr>
<tr>
<td></td>
<td>with high</td>
<td>generating catalyst</td>
<td>generating catalyst</td>
</tr>
<tr>
<td>FFAs</td>
<td>many times</td>
<td>many times</td>
<td>(from hours to</td>
</tr>
<tr>
<td>(&quot;/&gt;2 wt %)</td>
<td>times</td>
<td>times</td>
<td>minutes)</td>
</tr>
<tr>
<td>No pretreatment</td>
<td>Operates at mild temperature</td>
<td>Simultaneous</td>
<td>Saves cost of</td>
</tr>
<tr>
<td>required</td>
<td>(50 to 80 °C)</td>
<td>transesterification</td>
<td>purchasing catalyst</td>
</tr>
<tr>
<td></td>
<td></td>
<td>of TGs and esterifi-</td>
<td>activity and select-</td>
</tr>
<tr>
<td></td>
<td></td>
<td>cation of FA</td>
<td>tivity</td>
</tr>
<tr>
<td>Lower alcohol-</td>
<td>Simpler and less energy inten-</td>
<td>Simpler and less en-</td>
<td>Minimizes energy</td>
</tr>
<tr>
<td>to-oil (S:1) molar ratio</td>
<td>sive</td>
<td>ergy intensive</td>
<td>consumption</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>High biodiesel</td>
<td>Does not require feedstock</td>
<td>Easy separation of</td>
<td>Eliminates saponifi-</td>
</tr>
<tr>
<td>yield</td>
<td>pre-treatment</td>
<td>products</td>
<td>cation</td>
</tr>
<tr>
<td>Catalysts are</td>
<td>Mild reaction conditions &amp; less</td>
<td>Mild reaction con-</td>
<td>Reduces reaction</td>
</tr>
<tr>
<td>cheap and readily</td>
<td>prone to leaching</td>
<td>ditions &amp; less</td>
<td>length-to-diameter</td>
</tr>
<tr>
<td>available</td>
<td></td>
<td>prone to leaching</td>
<td>ratio</td>
</tr>
<tr>
<td>Waste generation</td>
<td>Minimized</td>
<td>Mild reaction</td>
<td>Short reaction time</td>
</tr>
<tr>
<td></td>
<td></td>
<td>conditions</td>
<td></td>
</tr>
<tr>
<td>Relative fast re-</td>
<td>Enhances mass transfer</td>
<td></td>
<td></td>
</tr>
<tr>
<td>action rates</td>
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<td></td>
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<tr>
<td>Easy product sep-</td>
<td>Saves cost of pur-</td>
<td></td>
<td></td>
</tr>
<tr>
<td>aration</td>
<td>chasing catalysts</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Minimizes solva-</td>
<td>by action of water</td>
<td></td>
<td></td>
</tr>
<tr>
<td>tion of active</td>
<td>Eliminates of saponifi-</td>
<td></td>
<td></td>
</tr>
<tr>
<td>sites</td>
<td>cation</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Chemical catalysed</td>
<td>Chemical catalysed (Modified)</td>
<td>Biochemical catalysed</td>
<td>Noncatalysed</td>
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<td>-------------------</td>
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<tr>
<td>Homogeneous acid</td>
<td>Homogeneous base</td>
<td>Heterogeneous acid</td>
<td>Heterogeneous base</td>
</tr>
<tr>
<td>Reduces size &amp; cost of reaction vessel</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Very attractive commercially</td>
<td></td>
<td></td>
<td></td>
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</tbody>
</table>

<table>
<thead>
<tr>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
<th>Challenges</th>
</tr>
</thead>
<tbody>
<tr>
<td>Very slow reaction rate &amp; mineral acids used are corrosive to the equipment</td>
<td>Availability of specific catalysts at low cost. Researches are ongoing to find low cost precursors.</td>
<td>Requires feedstock pretreatment &amp; catalyst get poisoned with prolong exposure to ambient air</td>
<td>Difficulties in process scale-up from laboratory scale to large-scale</td>
<td>Difficulties in process scale-up from laboratory scale to large-scale</td>
<td>Difficulties in process scale-up from laboratory scale to large-scale</td>
<td>High cost of enzymes</td>
<td>Energy intensive</td>
<td></td>
</tr>
<tr>
<td>Catalyst required in large quantities</td>
<td>Requires refined feedstock (0.5% FFA; 0.06% H₂O)</td>
<td>Limitation due to diffusion problems. This is solved by designing catalysts with large interconnected pores with high concentration of acid sites</td>
<td>High cost of reacting vessels</td>
<td>Depth of radiation is limited to a few cm.</td>
<td>Requires advanced technology</td>
<td>Requires advanced technology</td>
<td>Requires advanced technology</td>
<td>High production cost</td>
</tr>
<tr>
<td>Requires high alcohol-to-oil molar ratio and FFAAs react with the catalyst</td>
<td>Water saponifies the esters and FFAAs react with the catalyst</td>
<td>Two-step reaction of esterification and transesterification</td>
<td>Requires advanced technology</td>
<td>Safety issues in equipment handling</td>
<td>Safety issues in equipment handling</td>
<td>Enzymes</td>
<td>Not commercially profitable</td>
<td></td>
</tr>
<tr>
<td>Higher temperature</td>
<td>Requires methanol-to-oil of 6:1 (or higher) molar ratio instead of the stoichiometric 3:1 ratio</td>
<td>Product contamination from leaching of active catalytic sites</td>
<td>Safety issues in equipment handling.</td>
<td>Not commercially profitable</td>
<td>Safety issues</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Undesirable etherification reaction</td>
<td>Soap formation (FFA+/-&gt;2%)</td>
<td>Water saponifies the esters and FFAAs react with the catalyst making purification difficult</td>
<td>Very slow reaction rates (slower than homogeneous acid catalysed)</td>
<td>High temperature and pressure</td>
<td></td>
<td></td>
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<td></td>
</tr>
<tr>
<td>Chemical catalysed</td>
<td>Chemical catalysed (Modified)</td>
<td>Biochemical catalysed</td>
<td>Noncatalysed</td>
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</tr>
<tr>
<td>Homogeneous acid</td>
<td>Homogeneous base</td>
<td>Heterogeneous acid</td>
<td>Heterogeneous base</td>
<td>Microwave irradiation</td>
<td>Ultrasound (sonication)</td>
<td>Oscillatory flow reactor</td>
<td>Enzyme</td>
<td>Supercritical methanol</td>
</tr>
<tr>
<td>Separation and purification of glycerol</td>
<td>Loss of catalyst</td>
<td>Catalyst leaching leads to product contamination</td>
<td>Sensitive to methanol</td>
<td>Energy intensive</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Not commercially profitable</td>
<td>Reduces biodiesel yield &amp; generates wastewater</td>
<td>Purification decreases biodiesel yield</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Table 5. Merits and challenges surrounding transesterification processes[8]

6. Current challenges and future prospects

In order to make biodiesel profitable, several technical challenges need to be resolved. The most important challenge is in reducing the high cost of feedstock. Low-cost feedstocks such as algal oils, used cooking oils and animal fats are utilized to increase biodiesel profitability. However, presence of higher amounts of water and FFAs in these feedstocks poses the problems of saponification and extra pretreatment and purification costs with alkali catalysts. The challenge facing researchers currently is developing efficient heterogeneous acid catalysts that would alleviate these problems. Also, diversifying the by-product of biodiesel production processes is critical to ensuring its economic, social and environmental sustainability.

6.1. Vegetable oil as feedstock for biodiesel

Currently, biodiesel production costs are higher than those of petroleum diesel. Subsidies such as tax exempt and excise duty reductions are essential to make biodiesel price-competitive. It is not certain whether these political supports will be sustained in the future. It is therefore crucial for the biofuel industry to establish readily available and affordable feedstocks and efficient production systems to sustain its market growth.

6.2. Non-food crops

Early studies have indicated relative differences in the cultivation patterns and oil production management of the non-food feedstocks compared to food crops. These are still under investigation [79]. Therefore, more data is needed to evaluate the sustainability index to estimate the real global impact of these feedstocks. Microalgae are promising in solving most of the problems associated with energy crops. However, the cultivation and extraction technologies are still at their infancy and need major advancements for sustainable commercial production [1]. The oil extraction methods currently in use for algal
oil are expensive. Efficient mixing from pumps or motionless mixers is required to ensure homogeneity and to reduce mass transfer limitations. However, this increases the dispersion of glycerol into the FAME phase and the time required for separation. Techniques that utilize motionless mixing requires higher temperature and pressure to achieve shorter residence time. This increases energy consumption and cost implications. This aspects of biodiesel production technology is still being developed.

6.3. Effects of moisture and FFA

The key parameters that determine the viability of most feedstocks is FFA and moisture content. Pretreatments to less than 0.05% FFA is required for homogeneous alkali catalysts [80,81]. Prolonged storage in the presence of water and air leads to microbial growth and fuel degradation. This contributes to deposit formation on fuel injectors and engine damage. Heterogeneous acid catalysts are utilized to avoid the pretreatment and post production costs and storage problems.

6.4. Pyrolysis

Pyrolysis generates aromatic toxins. The bio-oil produced is corrosive due to high acidity, water content and other impurities such as solids and salts. These and other problems such as variable viscosity make it unstable and unsuitable for direct use [82]. It has 40% less energy density compared to diesel fuel because of the high oxygen content [83]. Depending on the feedstock and reacting conditions used, bio-oil is 10 to 100% more expensive than petroleum diesel. There is also the need to establish standards for product quality, use and distribution [1]. In order to stabilize the composition of the bio-oil and reduce water and oxygen content, processes such as steam reforming, hydro-treatment, hydro-cracking and emulsification with mineral diesel for direct use are employed [84-87].

6.5. Alcohol

Methanol is toxic, highly flammable and contributes to global warming. Gaskets and rubber seals made from natural rubber get easily deteriorated when biodiesel containing a high level of alcohol is used [80]. Therefore, control or replacement of the alcohol content is required. The biodiesel produced with methanol from fossil sources has approximately 94 to 96% biogenic content. In order to produce a 100% renewable biodiesel (fatty acid ethyl ester; FAEE), bioethanol is currently experimented as a substitute for methanol [88]. However, it is expensive to purify and recover ethanol because it forms an azeotrope with water. Additionally, chemical grade ethanol is usually denatured with poisonous substances to prevent it from being abused. Therefore, it is difficult to obtain pure chemical grade ethanol.

6.6. Supercritical alcohol process

The residence time for this process is within 4 s to 10 min because of efficient mixing [71,72]. However, due to higher reacting conditions of temperature and pressure, the process is faced with some limitations. Process scale-up for commercial production is the major one amongst
them. The process requires more energy at extra cost and higher molar ratio of alcohol-to-oil (42:1). Also, there is the need to quench the reaction in a rapid manner. This prevents the biodiesel from decomposing as a result of the high temperature and pressure. To reduce the high operating conditions and increase product yield, some researchers employ co-solvents, such as hexane, CO₂, and CaO [72]. Oil and alcohol are sparingly soluble in each other. However, small amount of hexane (2.5 wt%) added increased the biodiesel yield from 67.7% to 85.5% under supercritical conditions [72]. This was made possible because the co-solvent increased the homogeneity of the reactants. Supercritical CO₂ is a facile substance that can obtained at affordable cost. It is also environmentally friendly and can effectively used in the reaction and safely recovered via depressurization. A process that combines co-solvents in supercritical conditions is promising in increasing product yield, reducing process time and overall production costs.

6.7. Biodiesel/glycerol separation and FAME quality

The slightly soluble nature of FAMEs and glycerol makes product separation a necessary step. The product is usually allowed to settle for some hours into the different phases. However, the solubility of glycerol in ester and vice versa is increased in the presence of excess unreacted methanol which acts as solvent. This solvent action by the methanol increases the post production costs. Besides, it is also essential to remove all traces of TGs which form emulsion layer between the two phases. The presence of this layer further makes separation difficult and expensive. On the other hand, the storage, transportation, distribution and retail infrastructure used for petroleum diesel can be used for biodiesel even in its neat form. This will reduce construction costs for establishing new insfrastructures for biodiesel. However, biodiesel degrades after long period of storage. In order to prevent this from occuring, advances in storage and distribution logistics have to be developed. Also, similar logistics employed by the petroleum industry could be adapted.

6.8. Use of cosolvents

A technique developed to overcome mass transfer limitations and to increase the rate of reaction is the use of cosolvents such as methyl tert-butyl ether (MTBE) and tetrahydrofuran (THF). High quality FAMEs is obtained at moderate conditions (30 °C) within 10 minutes. However, the process requires larger and special “leak proof” reacting vessels and complete removal of the cosolvent from the product.

6.9. NOₓ emissions

Despite the favorable environmental impacts in terms of overall reduced GHG emissions, biodiesel has the potential to increase NOₓ emissions. Approximately 3 to 4%, 4 to 6% and 6 to 9% over petroleum diesel is emitted from B20, B40 and B100 respectively [89]. Adjustments in combustion temperatures and injection timing [90], use of antioxidants [91] and catalytic conversion techniques were successful in reducing these emissions [90].
6.10. Economic analysis

As discussed in the introductory section, vegetable oils have other important uses. Recently, dielectric oils and synthetic lubricants used for electric transformers have joined the market competition for these raw materials. This will impact negatively towards the cost of raw materials for the biodiesel industry [92]. About 15% of lubricants used in vehicles in some European countries are from vegetable oil derivatives [93]. Additionally, the heating value of biodiesel is 10% lower than that of petroleum diesel. This is because of the substantial amount of oxygen in the fuel. Moreover, it also has a higher specific gravity of 0.88 when compared to 0.85 of petroleum diesel. Therefore, its overall energy content per unit volume is having an impact which is approximately 5% lower than that obtained from petroleum diesel [94]. This results in higher specific fuel consumption values of the biodiesel. Another problem encountered when switching from petroleum diesel to biodiesel in the same fuel system is the clogging of the fuel filters. This is because biodiesel acts as a solvent which dissolves sediments in diesel fuel tanks [95]. On a positive note, sales of purified glycerol (glycerine) saved 6.5% of the operational cost [97-99] while 25% saving was reported in ref [96] from the utilization of waste soapstock with respect to virgin soybean oil. However, it is necessary to compensate the negative cost implications from commercial production of biodiesel from such low value feedstocks before valid conclusions can be derived.

7. Conclusions

Some of the major challenges faced by the biodiesel industry include readily available and affordable feedstocks, competition from a popular and cheaper energy source, technological advancements and acceptability. Those challenges requiring immediate attention are product stability under long storage, lower energy content, cold flow properties, catalyst leaching, microalgal oil extraction and NOx emissions. Despite these challenges however, the historical development of biodiesel is intriguing. Biodiesel has successfully remained an energy source to be reckoned with even after being relegated to the background for so many years. Concerns over diminishing oil reserves, increasing crude oil prices and associated environmental impacts aided the reemergence of biodiesel; making it the fastest growing industry worldwide. Several technologies were developed while more advances are in the process of being established. Other successes associated with the biodiesel industry include reduction in environmental impacts, job creation, energy security and waste-utilization. Biodiesel is regarded as a viable alternative or additive to petrodiesel because of its good properties such as nontoxicity, clean-burning, renewability and acceptability. Consequently, the prospects of the biodiesel industry are numerous. The biodiesel production process is shifting from other sources to algal oil and heterogeneous acid catalysts. Algal oil is a more reliable and efficient source. It has the potential of producing yields of more than 100 times those attainable per hectare from oilseeds. Affordable and readily available non-food feedstocks such as microalgae have been produced in commercial scale without competing with arable land or causing deforestation. Additionally, the use of heterogeneous acid catalyst produces cleaner and higher yields.
It employs cheaper and readily available feedstocks and minimizes pre- and post-product costs. These and other factors such as waste-utilization and cleaner emissions will help ensure biodiesel as a cheaper energy source with greater economic benefits and healthier environments.

**Author details**

Y.M. Sani\textsuperscript{1,2}, W.M.A.W. Daud\textsuperscript{1}” and A.R. Abdul Aziz\textsuperscript{1}

*Address all correspondence to: ashri@um.edu.my

1 Department of Chemical Engineering, Faculty of Engineering, University Malaya, Kuala Lumpur, Malaysia

2 Department of Chemical Engineering, Ahmadu Bello University, Zaria, Nigeria

**References**


